

# Predicted Structures and Spectroscopic Characteristics of Hydrazine, Lithium-substituted Hydrazine and Their Higher Derivatives

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## Calculation method.

Density functional theory, B3LYP hybrid exchange-correlation functional is employed to investigate a potential of molecules as novel propellant..

Dunning's aug-cc-pVDZ (5d) basis set is used for H, N, C, F elements, while 6-31G(d) basis set is used for Li atom, where aug-cc-pVDZ is not available.

Heat of formation is calculated as energy difference between a target molecule and constituent atoms in their standard state. Since H<sub>2</sub>, O<sub>2</sub>, F<sub>2</sub> are gas in their standard state,

Calculated value are used without modification, however standard state of C and Li are solid, thus experimental heat of formation of each atom 171.29 kcal/mol and 38.1 kcal/mol are used together with calculated total enthalpy.

## Notation.

In the pictures of molecules, N,C,H and Li are given by skyblue, darkgreen, grey and brown, respectively.

Bond length between N,C and Li are given in Angstrom. Full molecular coordinate are available by request.

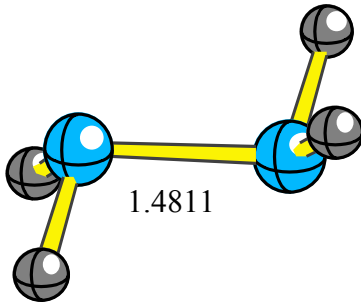
H<sub>f</sub> denotes heat of formation and I<sub>sp</sub> denotes specific impulse.

## Investigation of HEDM molecules.

### *Propellants which are currently in use.*

$N_2H_4$ , Hydrazine ( $1_i$ )

$C_{2h}$ ,  $^1A_g$ ,  $H_f=15.7$  kcal/mol,  $I_{sp}=185.7$  sec.

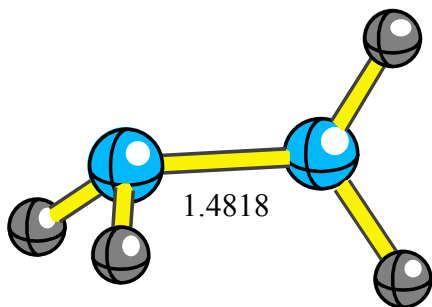


No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	Au	-25.8	110.3	0.0	0.65
2	Ag	945.9	0.0	13.7	0.29
3	Bu	1034.4	160.8	0.0	0.61
4	Au	1104.7	1.1	0.0	0.61
5	Ag	1214.5	0.0	4.0	0.06
6	Bg	1475.7	0.0	3.1	0.75
7	Bu	1609.5	37.3	0.0	0.41
8	Ag	1671.4	0.0	6.1	0.64
9	Ag	3428.7	0.0	254.8	0.10
10	Bu	3446.2	2.6	0.0	0.51
11	Bg	3499.6	0.0	104.3	0.75
12	Au	3518.3	2.2	0.0	0.70

2. Dissociation into  $2NH_2$ .

N<sub>2</sub>H<sub>4</sub>, Hydrazine

C<sub>2</sub>, <sup>1</sup>A

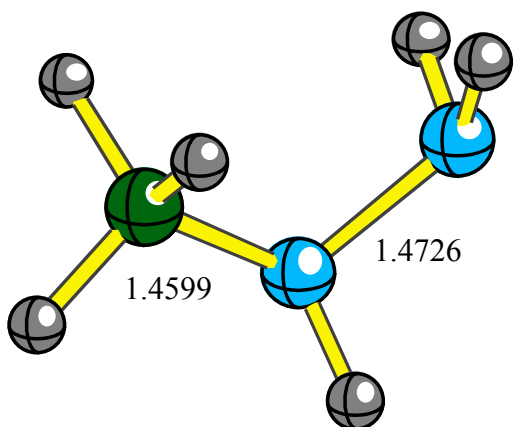


No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	439.0	35.8	0.6	0.65
2	A	801.3	60.9	0.2	0.73
3	B	970.8	122.4	0.6	0.75
4	A	1112.7	11.3	11.2	0.11
5	B	1284.8	5.0	0.8	0.75
6	A	1320.2	3.0	1.2	0.72
7	B	1649.5	11.2	4.3	0.75
8	A	1665.9	8.5	6.3	0.48
9	B	3442.5	13.5	62.8	0.75
10	A	3451.0	3.2	229.9	0.01
11	A	3549.3	1.7	98.3	0.67
12	B	3554.1	2.8	21.8	0.75

2. Dissociation into 2NH<sub>2</sub>.

CN<sub>2</sub>H<sub>6</sub>, Monomethylhydrazine (MMH)

C<sub>1</sub>, <sup>1</sup>A, H<sub>f</sub>=15.70 kcal/mol, I<sub>sp</sub>=154.7 sec

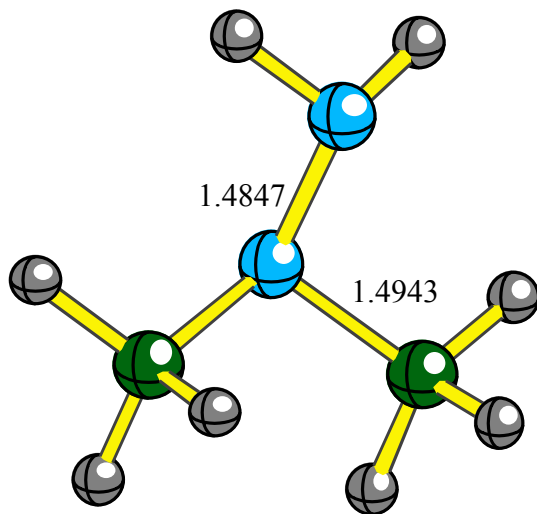


No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	263.8	3.0	0.3	0.68
2	A	343.9	24.5	0.9	0.37
3	A	421.4	6.5	1.5	0.25
4	A	780.4	99.3	2.3	0.59
5	A	911.2	64.5	0.3	0.66
6	A	980.3	26.5	6.6	0.27
7	A	1124.8	6.7	1.8	0.46
8	A	1138.2	9.8	2.3	0.70
9	A	1215.2	4.1	3.3	0.29
10	A	1304.7	6.4	1.2	0.66
11	A	1423.0	0.3	2.1	0.75
12	A	1454.8	9.4	6.0	0.75
13	A	1470.2	3.0	3.7	0.71
14	A	1506.3	3.7	3.3	0.73
15	A	1659.5	13.0	5.2	0.63
16	A	2936.0	90.9	170.7	0.11
17	A	3056.7	37.4	116.1	0.23
18	A	3103.1	27.1	83.9	0.64
19	A	3378.6	18.6	156.8	0.11
20	A	3515.6	2.8	136.4	0.26
21	A	3532.3	0.9	28.3	0.75

8. Dissociation into CNH<sub>4</sub> + NH<sub>2</sub> or CH<sub>3</sub> + N<sub>2</sub>H<sub>3</sub>.

$C_2N_2H_8$ , 1,1-Dimethyl hydrazine (UDMH)

$C_s$ ,  $^1A'$ ,  $H_f=20.16$  kcal/mol,  $I_{sp}=153.5$  sec



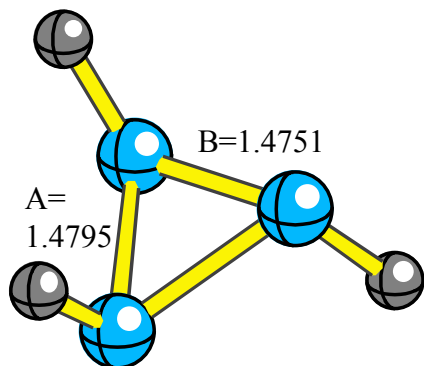
No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A''	154.5	35.0	0.2	0.75
2	A''	260.5	8.8	0.0	0.75
3	A'	277.9	0.1	0.9	0.72
4	A'	412.7	4.4	1.6	0.34
5	A''	438.4	1.6	1.3	0.75
6	A'	441.5	18.1	1.9	0.23
7	A'	818.6	15.0	13.1	0.37
8	A'	983.6	20.2	9.5	0.35
9	A''	1035.4	11.9	1.5	0.75
10	A''	1092.5	1.0	0.9	0.75
11	A'	1100.5	46.2	3.7	0.16
12	A''	1127.5	10.9	2.6	0.75
13	A'	1183.8	3.7	8.4	0.32
14	A'	1247.3	0.2	2.1	0.66
15	A''	1389.9	3.8	0.0	0.75
16	A''	1408.2	1.2	2.1	0.75
17	A'	1435.2	3.6	2.6	0.62
18	A''	1451.6	9.5	5.6	0.75
19	A'	1462.3	8.3	8.5	0.72
20	A''	1471.5	1.2	5.2	0.75
21	A'	1485.3	14.0	3.5	0.75
22	A'	1632.8	35.7	3.5	0.75
23	A''	2947.8	50.7	24.1	0.75
24	A'	2956.4	103.7	281.6	0.06
25	A''	3066.2	35.6	52.7	0.75
26	A'	3068.6	31.8	143.9	0.21
27	A''	3113.5	7.4	20.2	0.75
28	A'	3118.7	39.3	106.0	0.66
29	A'	3422.3	2.0	149.6	0.12
30	A''	3502.4	0.1	66.6	0.75

7. Dissociation into  $NH_2 + C_2NH_6$ .

**Higher derivatives of hydrazine:  $N_xH_y$  molecules.**

$N_3H_3$ , Cyclotriazane (c,t)

$C_s$ ,  $^1A'$ ,  $H_f=85.66$  kcal/mol,  $I_{sp}=365.4$  sec.



No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A''	787.0	9.9	6.7	0.75
2	A'	790.7	8.1	6.8	0.74
3	A''	1099.9	28.6	2.3	0.75
4	A'	1121.8	55.0	2.0	0.39
5	A'	1166.4	17.7	22.9	0.08
6	A'	1281.1	42.7	5.4	0.32
7	A''	1288.5	41.7	0.2	0.75
8	A''	1424.5	12.8	4.9	0.75
9	A'	1459.6	0.3	13.2	0.10
10	A''	3371.2	0.0	68.9	0.75
11	A'	3398.0	1.0	197.0	0.14
12	A'	3426.1	1.4	53.1	0.37

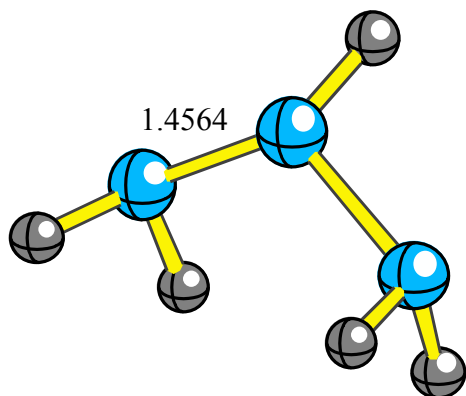
1. Open ring at A.

2. Dissociation into NH +  $N_2H_2$  (breaks two bonds A and B).

5. Dissociation into 3NH (breaks three bonds).

$N_3H_5$ , Triazine

$C_s$ ,  $^1A'$ ,  $H_f=32.2$  kcal/mol,  $I_{sp}=219.1$  sec.

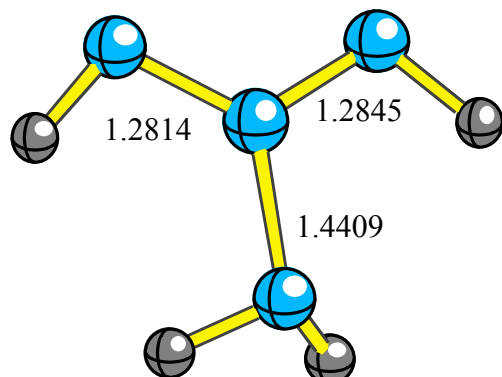


No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A''	333.9	0.1	0.4	0.75
2	A'	437.1	52.2	1.8	0.20
3	A'	477.0	0.1	4.9	0.28
4	A'	785.1	45.4	3.6	0.40
5	A''	830.5	265.8	0.6	0.75
6	A'	956.6	28.8	2.6	0.50
7	A''	1091.0	0.8	1.2	0.75
8	A'	1154.1	21.4	10.4	0.33
9	A''	1230.1	3.5	1.8	0.75
10	A'	1344.5	11.3	0.3	0.25
11	A''	1507.8	0.1	1.7	0.75
12	A''	1640.6	0.5	6.0	0.75
13	A'	1666.7	27.9	3.7	0.55
14	A'	3367.6	30.5	278.9	0.10
15	A''	3380.5	2.4	18.5	0.75
16	A'	3516.3	0.3	179.6	0.32
17	A''	3532.6	1.0	22.7	0.75
18	A'	3545.3	6.2	16.0	0.41

7. Dissociation into  $NH_2 + N_2H_3$ .

$N_4H_4$ , 2-aminotriazene

$C_1$ ,  $^1A$ ,  $H_f=80.14$  kcal/mol,  $I_{sp}=306.1$  sec.



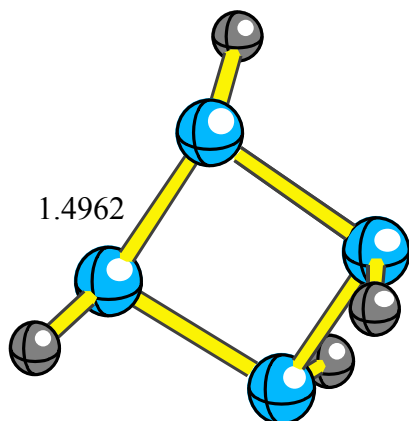
This structure is preferred over a structure with a three-member nitrogen ring structure.

No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	103.8	32.3	0.2	0.69
2	A	506.7	1.6	0.6	0.30
3	A	563.9	5.7	3.1	0.74
4	A	618.7	10.8	4.6	0.26
5	A	655.2	27.6	1.9	0.34
6	A	778.4	92.5	0.3	0.11
7	A	810.0	130.8	3.6	0.16
8	A	990.0	58.3	17.8	0.11
9	A	1227.2	76.6	16.9	0.32
10	A	1256.4	54.4	33.0	0.23
11	A	1302.3	291.1	0.0	0.74
12	A	1494.5	85.7	4.8	0.06
13	A	1624.4	42.4	4.7	0.44
14	A	1690.3	132.2	5.4	0.72
15	A	3462.2	8.5	85.3	0.09
16	A	3466.9	16.4	35.2	0.75
17	A	3478.2	2.2	158.2	0.15
18	A	3568.2	15.7	42.7	0.75

7. Dissociation into  $NH_2 + N_3H_2$ .

$N_4H_4$  , Cyclotetrazane

$C_2$ ,  $^1A$ ,  $H_f=108.95$  kcal/mol,  $I_{sp}=356.9$  sec.



No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	328.4	0.0	2.0	0.70
2	B	846.4	1.1	0.0	0.75
3	B	846.4	1.1	0.0	0.75
4	A	890.8	1.3	2.8	0.75
5	A	969.6	0.0	14.7	0.13
6	A	1024.0	0.0	10.1	0.75
7	A	1091.9	2.4	3.3	0.75
8	B	1119.2	153.1	0.7	0.75
9	B	1119.2	153.1	0.7	0.75
10	A	1271.6	0.0	0.0	0.07
11	A	1370.8	0.0	13.5	0.04
12	B	1451.1	2.3	4.9	0.75
13	B	1451.1	2.3	4.9	0.75
14	A	1491.0	0.0	0.8	0.75
15	A	3388.3	0.0	337.9	0.12
16	B	3414.9	0.3	56.4	0.75
17	B	3415.0	0.3	56.4	0.75
18	A	3429.1	0.1	6.6	0.75

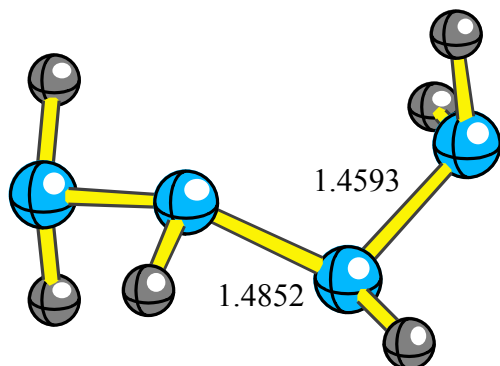
2,3. Dissociation into  $NH + N_3H_3$  (two bonds breaking).

5. Dissociation into  $4NH$ . (four bonds breaking).

6. Dissociation into  $2N_2H_2$  (two bonds breaking).

$N_4H_6$ , Tetrazine

$C_2$ ,  $^1A$ ,  $H_f=53.07$  kcal/mol,  $I_{sp}=245.04$  sec.



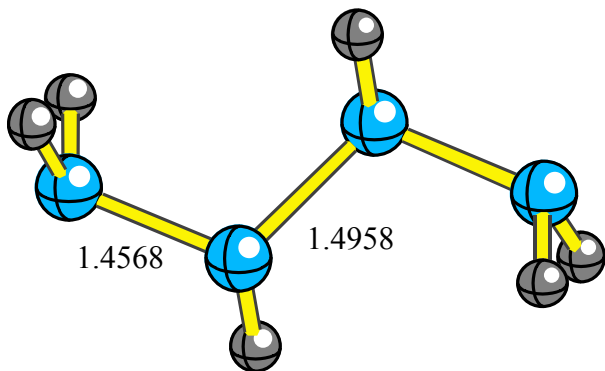
No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	99.0	2.4	1.1	0.72
2	B	340.0	49.0	0.0	0.75
3	A	362.3	31.4	0.5	0.74
4	A	469.8	1.2	8.4	0.15
5	B	550.5	81.6	0.6	0.75
6	A	783.6	3.0	5.1	0.75
7	B	867.2	161.6	0.8	0.75
8	A	881.9	15.0	0.5	0.69
9	B	978.4	159.6	1.8	0.75
10	A	994.6	2.1	9.1	0.20
11	A	1117.5	0.4	8.3	0.19
12	B	1200.2	10.4	1.1	0.75
13	A	1310.8	1.1	0.7	0.73
14	B	1324.9	19.2	0.5	0.75
15	A	1493.2	0.5	4.8	0.28
16	B	1506.7	0.3	3.0	0.75
17	A	1657.7	17.9	4.1	0.62
18	B	1659.8	11.2	5.7	0.75
19	B	3406.4	11.0	11.1	0.75
20	A	3409.2	3.7	308.3	0.07
21	B	3459.4	3.4	33.5	0.75
22	A	3479.4	0.1	204.1	0.21
23	B	3544.7	2.7	18.7	0.75
24	A	3545.7	2.7	59.5	0.64

5. Dissociation into  $2N_2H_3$ .

6. Dissociation into  $NH_2 + N_3H_4$ .

$N_4H_6$ , Tetrazine (t)

$C_i$ ,  $^1A_g$ ,  $H_f=54.02$ ,  $I_{sp}=247.2$  sec.



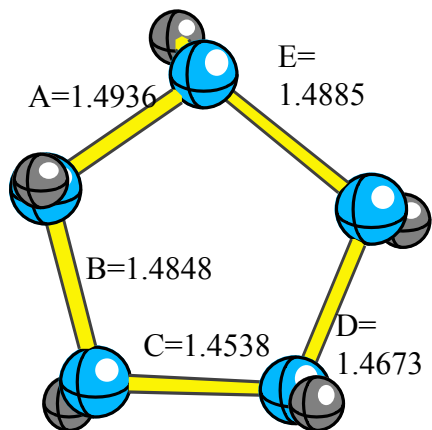
No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	Au	88.0	4.3	0.0	0.00
2	Au	310.8	25.0	0.0	0.00
3	Au	394.6	71.3	0.0	0.00
4	Ag	450.3	0.0	4.6	0.24
5	Ag	500.4	0.0	13.7	0.25
6	Ag	804.9	0.0	6.2	0.74
7	Au	880.1	275.8	0.0	0.00
8	Ag	909.9	0.0	12.7	0.52
9	Au	988.6	109.7	0.0	0.00
10	Ag	1073.7	0.0	3.9	0.75
11	Au	1112.1	23.5	0.0	0.00
12	Ag	1203.6	0.0	8.4	0.16
13	Au	1282.7	6.5	0.0	0.00
14	Ag	1351.7	0.0	2.1	0.17
15	Au	1471.4	7.4	0.0	0.00
16	Ag	1535.1	0.0	3.3	0.64
17	Au	1655.3	31.0	0.0	0.00
18	Ag	1655.6	0.0	10.1	0.62
19	Au	3397.6	20.8	0.0	0.00
20	Ag	3402.7	0.0	364.8	0.09
21	Ag	3439.7	0.0	151.5	0.36
22	Au	3459.4	4.5	0.0	0.00
23	Au	3542.7	8.0	0.0	0.00
24	Ag	3544.2	0.0	110.2	0.61

5. Dissociation into  $2N_2H_3$ .

11. Dissociation into  $NH_2 + N_3H_4$ .

# $N_5H_5$ , Cyclepentazane

$C_1$ ,  $^1A$ ,  $H_f=103.16$  kcal/mol,  $I_{sp}=310.6$  sec.

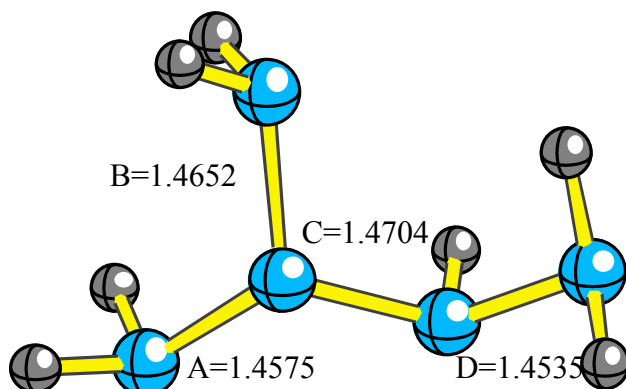


No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	155.7	2.3	0.1	0.75
2	A	459.4	0.1	1.2	0.57
3	A	707.8	9.7	7.4	0.34
4	A	778.7	12.1	2.9	0.75
5	A	833.2	28.1	8.7	0.09
6	A	861.4	71.7	0.1	0.74
7	A	867.5	54.8	0.4	0.27
8	A	890.7	13.9	8.3	0.09
9	A	921.5	16.6	0.1	0.75
10	A	1035.8	15.5	1.0	0.67
11	A	1038.7	27.0	3.0	0.75
12	A	1079.8	98.6	2.6	0.41
13	A	1213.4	61.5	0.5	0.75
14	A	1259.5	26.7	15.4	0.04
15	A	1368.4	6.8	0.6	0.75
16	A	1426.0	2.9	5.1	0.08
17	A	1463.4	4.4	4.8	0.75
18	A	1528.5	0.5	1.9	0.75
19	A	1532.2	0.1	7.6	0.71
20	A	3410.9	0.2	20.8	0.75
21	A	3441.5	0.7	296.5	0.14
22	A	3457.8	0.2	57.0	0.40
23	A	3482.3	0.8	24.5	0.75
24	A	3490.4	6.0	24.9	0.36

3. Open ring at D.
4. Open ring at B.
5. Dissociation into  $NH + N_4H_4$  (two bonds breaking at D and E).
6. Open ring at A or C.
7. Open ring at B or E.
8. Open ring at A or C.
9. Dissociation into  $N_2H_2 + N_3H_3$  (two bonds breaking at A and C).

$N_5H_7$ , 2-aminotetrazine

$C_1, ^1A, H_f=72.55$  kcal/mol,  $I_{sp}=257.08$  sec.



No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	124.7	3.8	0.2	0.69
2	A	230.6	22.2	1.0	0.53
3	A	298.0	26.1	0.8	0.72
4	A	330.0	3.0	0.8	0.54
5	A	423.9	33.8	0.7	0.24
6	A	472.0	8.6	2.3	0.74
7	A	526.2	26.1	3.1	0.13
8	A	580.9	26.2	8.9	0.23
9	A	760.5	35.3	9.6	0.29
10	A	812.4	3.5	1.6	0.56
11	A	845.4	183.2	0.9	0.73
12	A	935.9	132.2	3.2	0.26
13	A	1010.4	67.9	4.5	0.30
14	A	1066.2	7.2	5.9	0.64
15	A	1161.8	17.6	4.2	0.75
16	A	1189.0	19.6	8.6	0.14
17	A	1242.1	1.9	2.5	0.47
18	A	1298.5	6.5	2.9	0.75
19	A	1365.2	14.9	0.7	0.62
20	A	1508.8	6.4	5.1	0.67
21	A	1615.0	1.7	5.4	0.75
22	A	1652.2	15.6	4.6	0.59
23	A	1659.5	30.4	7.1	0.75
24	A	3344.8	36.2	308.0	0.09
25	A	3355.4	1.6	26.0	0.73
26	A	3366.0	6.7	103.8	0.19
27	A	3409.8	9.1	87.2	0.10
28	A	3535.3	3.5	105.9	0.34
29	A	3540.2	4.8	66.2	0.55
30	A	3546.8	4.1	72.8	0.40

5. Dissociation into  $NH_2 + N_2H_2 + N_2H_3$  (two bonds breaking at B and C).

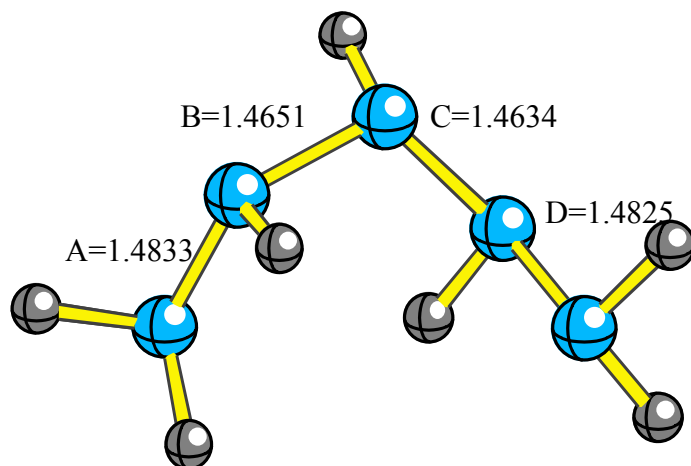
12. Dissociation into  $N_4H_6 + NH_2$  (bond breaking at D).

14. Dissociation into  $NH_2 + N_4H_6$  (bond breaking at B).

15. Dissociation into  $N_3H_4 + N_2H_3$  (bond breaking at C).

$N_5H_7$ , Pentazine (A)

$C_{1v}$ ,  $^1A$ ,  $H_f=74.58$  kcal/mol,  $I_{sp}=260.7$  sec.



No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	112.3	3.2	1.6	0.64
2	A	146.2	40.3	0.3	0.30
3	A	219.8	23.7	0.3	0.73
4	A	297.3	23.4	0.4	0.32
5	A	346.6	5.5	1.8	0.14
6	A	536.0	14.9	0.4	0.63
7	A	655.5	20.8	2.3	0.70
8	A	757.8	57.8	7.5	0.19
9	A	798.1	44.8	1.6	0.20
10	A	861.8	141.9	0.9	0.68
11	A	889.4	11.1	10.5	0.29
12	A	1021.6	8.0	2.1	0.24
13	A	1059.7	5.4	8.6	0.20
14	A	1096.7	126.9	1.0	0.74
15	A	1151.8	8.1	3.5	0.24
16	A	1172.2	26.6	6.8	0.26
17	A	1283.2	5.5	1.3	0.29
18	A	1332.0	8.6	2.8	0.50
19	A	1448.5	2.0	1.9	0.31
20	A	1522.7	3.2	3.5	0.67
21	A	1544.7	0.4	3.4	0.71
22	A	1619.5	27.6	4.3	0.73
23	A	1639.0	15.4	5.5	0.63
24	A	3359.1	15.0	64.4	0.16
25	A	3435.1	0.6	148.5	0.12
26	A	3462.0	1.3	135.9	0.12
27	A	3511.7	1.0	133.3	0.08
28	A	3517.8	2.4	67.7	0.69
29	A	3534.9	8.9	37.1	0.46
30	A	3564.1	8.6	71.3	0.67

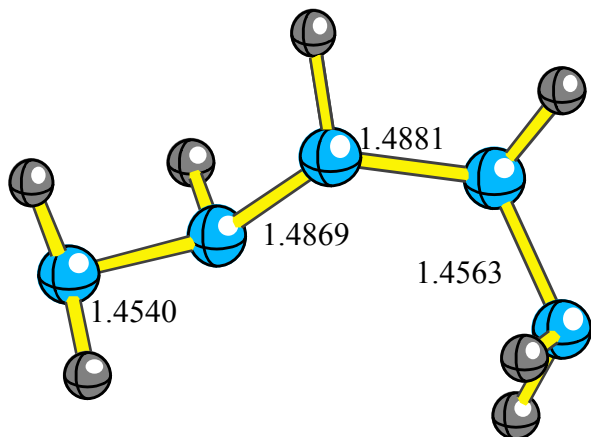
8. Dissociation into  $N_2H_3 + N_3H_4$  (bond breaking at B).

11. Dissociation into  $N_4H_5 + NH_2$  (bond breaking at D).

13. Dissociation into  $NH_2 + N_4H_5$  (bond breaking at A).

$N_5H_7$  , Pentazine (B)

$C_1$ ,  $^1A$ ,  $H_f=76.71$  kcal/mol,  $I_{sp}=264.4$  sec.

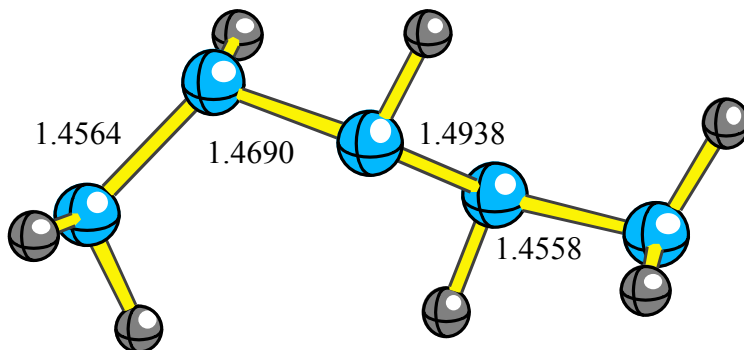


No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	60.1	0.4	0.8	0.73
2	A	126.3	7.2	0.6	0.55
3	A	292.4	11.6	0.2	0.05
4	A	353.9	25.3	1.4	0.21
5	A	413.6	44.8	1.5	0.64
6	A	425.6	17.0	6.6	0.18
7	A	573.0	54.6	3.5	0.51
8	A	772.0	33.9	3.1	0.62
9	A	850.0	11.6	5.3	0.75
10	A	908.0	193.8	0.4	0.72
11	A	917.9	41.0	3.4	0.74
12	A	990.1	56.1	4.7	0.18
13	A	1033.3	14.2	11.0	0.10
14	A	1066.1	33.1	3.1	0.29
15	A	1180.5	13.6	2.6	0.62
16	A	1203.5	18.2	3.2	0.39
17	A	1307.3	5.9	3.1	0.69
18	A	1327.2	10.6	0.8	0.69
19	A	1462.3	2.1	5.3	0.40
20	A	1486.3	0.3	4.1	0.65
21	A	1510.2	0.2	3.0	0.56
22	A	1666.3	14.5	4.3	0.60
23	A	1669.8	14.7	6.0	0.61
24	A	3338.9	16.7	193.7	0.21
25	A	3396.5	9.7	124.4	0.13
26	A	3407.4	8.8	153.2	0.11
27	A	3460.7	0.8	141.9	0.25
28	A	3474.1	3.2	144.5	0.20
29	A	3536.1	1.6	48.0	0.58
30	A	3546.6	5.4	31.8	0.59

No frequency corresponds to N-N bond breaking.

$N_5H_7$ , Pentazine (C)

$C_1$ ,  $^1A$ ,  $H_f=52.27$  kcal/mol,  $I_{sp}=218.2$  sec.

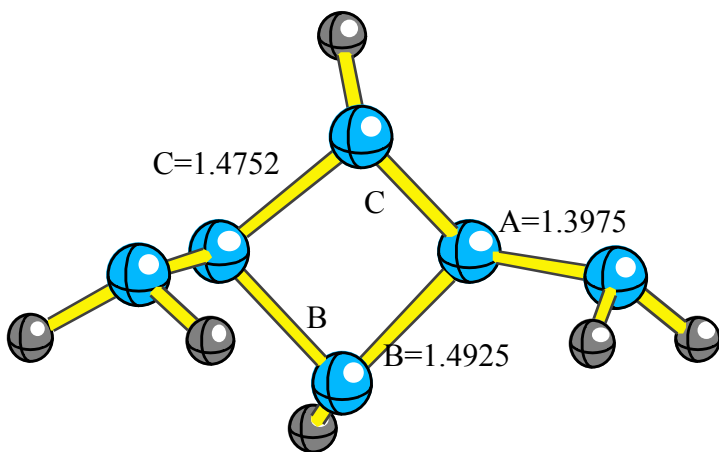


No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	45.9	2.3	0.4	0.72
2	A	143.7	8.4	0.7	0.69
3	A	253.4	13.8	0.2	0.19
4	A	362.6	23.0	4.0	0.13
5	A	405.2	32.0	2.4	0.42
6	A	449.5	20.1	7.9	0.19
7	A	555.4	53.3	4.2	0.42
8	A	769.7	12.3	4.7	0.72
9	A	831.6	122.5	4.0	0.73
10	A	888.1	91.7	7.1	0.54
11	A	908.3	101.3	2.5	0.72
12	A	973.6	51.4	2.8	0.23
13	A	1058.9	15.7	3.4	0.40
14	A	1087.8	32.7	0.9	0.09
15	A	1156.5	13.9	9.8	0.10
16	A	1220.3	7.3	1.9	0.25
17	A	1305.9	1.3	0.5	0.72
18	A	1318.2	7.5	1.7	0.43
19	A	1450.1	10.0	0.3	0.71
20	A	1497.0	1.0	0.9	0.73
21	A	1533.2	1.3	4.1	0.74
22	A	1654.9	15.4	4.3	0.53
23	A	1666.2	14.3	6.2	0.69
24	A	3402.2	9.9	149.2	0.11
25	A	3419.0	7.9	144.5	0.04
26	A	3437.7	1.4	143.6	0.06
27	A	3441.0	3.8	104.4	0.62
28	A	3469.5	3.3	22.5	0.30
29	A	3543.7	5.4	55.3	0.65
30	A	3546.6	3.1	57.3	0.51

No frequency corresponds to N-N bond breaking.

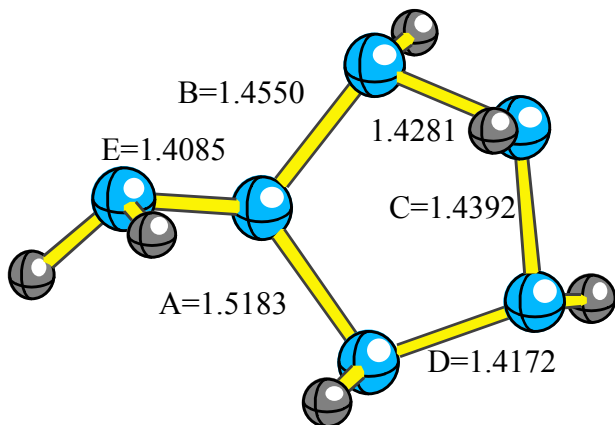
$N_6H_6$ , Paradiaminocyclotetrazane

$C_1$ ,  ${}^1A$ ,  $H_f=135.86$  kcal/mol,  $I_{sp}=325.4$  sec.



$N_6H_6$ , Aminocyclopentazane

$C_1$ ,  ${}^1A$ ,  $H_f=121.65$  kcal/mol,  $I_{sp}=307.9$  sec.



Vibrational frequency for N<sub>6</sub>H<sub>6</sub>, paradiaminocyclotetrazane.

No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	118.1	0.1	2.6	0.38
2	A	281.3	29.7	0.6	0.75
3	A	318.3	38.5	2.7	0.54
4	A	402.7	11.5	2.4	0.25
5	A	468.9	99.8	2.9	0.75
6	A	520.8	0.8	3.5	0.19
7	A	546.5	2.0	0.5	0.75
8	A	704.7	12.6	11.2	0.26
9	A	796.4	68.3	0.3	0.75
10	A	814.3	132.0	4.3	0.75
11	A	829.2	105.6	7.3	0.11
12	A	907.0	9.3	17.6	0.20
13	A	937.6	14.2	10.5	0.75
14	A	960.5	60.0	0.8	0.33
15	A	1114.2	80.6	0.4	0.62
16	A	1130.3	32.2	2.9	0.75
17	A	1135.7	1.9	10.1	0.03
18	A	1252.0	6.9	6.2	0.05
19	A	1288.4	17.1	0.8	0.75
20	A	1319.7	0.6	0.7	0.75
21	A	1340.2	1.8	0.5	0.67
22	A	1389.6	10.3	0.6	0.75
23	A	1629.0	3.1	7.8	0.75
24	A	1637.3	47.2	3.3	0.58
25	A	3410.2	4.4	31.5	0.27
26	A	3410.7	6.8	288.1	0.06
27	A	3429.7	1.1	81.8	0.55
28	A	3440.3	5.1	126.1	0.21
29	A	3568.6	14.4	60.8	0.58
30	A	3568.7	5.0	107.1	0.41

5. Ring opening at B.

12. Dissociation into NH + N<sub>5</sub>H<sub>5</sub> (two bonds breaking at B).

13. Dissociation into 2N<sub>3</sub>H<sub>3</sub> (two bonds breaking at B and C).

14. Dissociation into NH + N<sub>5</sub>H<sub>5</sub> (two bonds breaking at C).

Vibrational frequencies for N<sub>6</sub>H<sub>6</sub>, aminocyclopentazane.

No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	105.2	0.9	0.4	0.38
2	A	276.0	36.5	0.7	0.74
3	A	408.5	10.3	0.7	0.71
4	A	476.8	7.2	6.6	0.20
5	A	555.1	8.3	1.3	0.51
6	A	674.6	16.5	9.2	0.46
7	A	733.4	14.0	4.9	0.48
8	A	771.1	34.7	11.8	0.07
9	A	810.6	20.4	2.9	0.44
10	A	856.0	35.1	1.9	0.12
11	A	868.3	57.4	2.5	0.51
12	A	897.9	124.9	0.8	0.38
13	A	943.2	9.3	1.0	0.71
14	A	1019.0	20.4	1.4	0.61
15	A	1033.0	58.1	2.2	0.54
16	A	1097.0	52.3	1.2	0.75
17	A	1204.9	37.3	1.1	0.74
18	A	1265.0	35.3	15.5	0.04
19	A	1308.2	3.9	2.0	0.11
20	A	1370.8	3.5	0.2	0.75
21	A	1439.6	4.9	3.5	0.13
22	A	1515.1	1.3	5.6	0.73
23	A	1526.9	0.7	3.4	0.66
24	A	1637.7	19.2	6.0	0.55
25	A	3350.8	21.5	234.4	0.13
26	A	3365.6	0.2	16.2	0.38
27	A	3460.1	0.7	173.8	0.16
28	A	3482.2	1.0	23.6	0.73
29	A	3492.9	6.3	45.1	0.37
30	A	3567.7	13.4	77.5	0.41

6. Ring opening at A.

7. Ring opening at A or B.

8. Dissociation into N<sub>2</sub>H<sub>2</sub> + N<sub>4</sub>H<sub>4</sub> (two bonds breaking at A and B).

10. Ring opening at C.

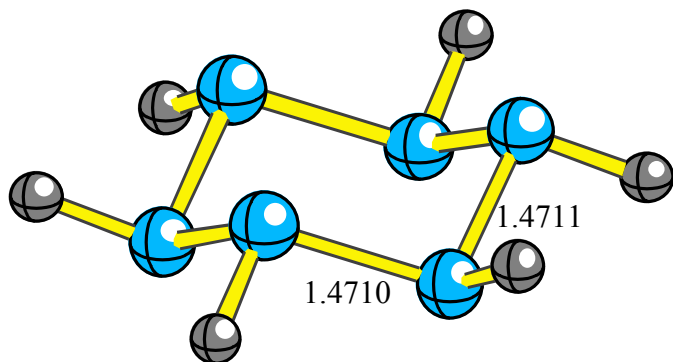
11. Ring opening at A,B, or C.

12. Ring opening at D.

13. Ring opening at E.

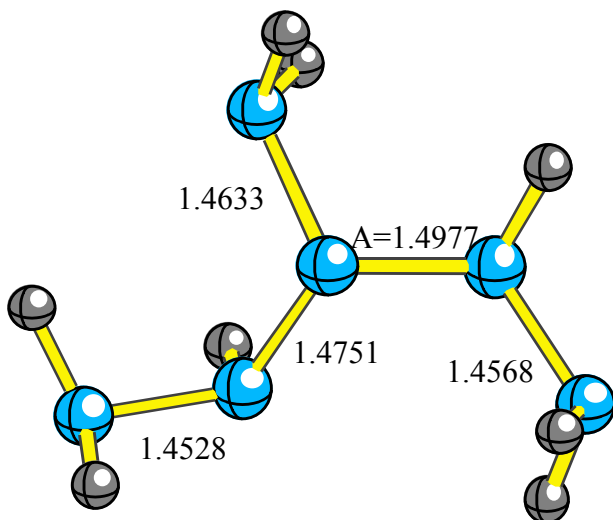
$N_6H_6$ , Cyclohexazane

$C_{2h}$ ,  $^1A_g$ ,  $H_f=153.20$  kcal/mol,  $I_{sp}=345.6$  sec.



$N_6H_8$ , 3-aminopentazine

$C_1$ ,  $^1A$ ,  $H_f=89.70$  kcal/mol,  $I_{sp}=261.5$  sec.



Vibrational frequencies of N<sub>6</sub>H<sub>6</sub>, cyclohexazane.

No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	Au	314.0	18.4	0.0	0.00
2	Bu	314.0	18.5	0.0	0.48
3	Ag	490.9	0.0	1.3	0.75
4	Bg	491.2	0.0	1.3	0.75
5	Ag	513.1	0.0	0.1	0.32
6	Bu	720.4	110.4	0.0	0.74
7	Au	843.5	50.8	0.0	0.73
8	Bu	843.5	50.8	0.0	0.75
9	Ag	843.6	0.0	5.5	0.73
10	Bg	843.9	0.0	5.4	0.75
11	Ag	845.7	0.0	29.5	0.17
12	Au	988.6	0.0	0.0	0.55
13	Au	1102.4	80.1	0.0	0.55
14	Bu	1102.6	80.1	0.0	0.74
15	Bg	1171.6	0.0	5.3	0.75
16	Ag	1171.7	0.0	5.4	0.75
17	Ag	1189.2	0.0	13.4	0.02
18	Bu	1241.4	83.8	0.0	0.58
19	Au	1353.6	0.0	0.0	0.62
20	Bg	1454.6	0.0	0.0	0.75
21	Ag	1468.9	0.0	4.7	0.75
22	Bg	1469.0	0.0	4.7	0.75
23	Bu	1498.0	38.5	0.0	0.60
24	Au	1498.2	38.5	0.0	0.21
25	Bu	3424.0	2.3	0.0	0.75
26	Bg	3428.0	0.0	107.6	0.75
27	Ag	3428.2	0.0	108.8	0.73
28	Au	3429.1	13.7	0.0	0.26
29	Bu	3429.4	13.7	0.0	0.75
30	Ag	3430.3	0.0	304.9	0.07

7. Open ring.

8. Dissociation into NH + N<sub>5</sub>H<sub>5</sub> (two bonds breaking).

12. Dissociation into 3N<sub>2</sub>H<sub>2</sub> (three bonds breaking).

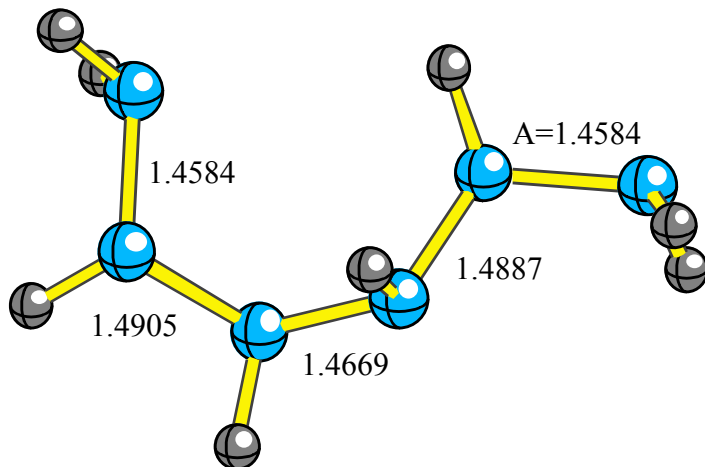
Vibrational frequencies of N<sub>6</sub>H<sub>8</sub>, 3-aminopentazine.

No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	120.8	2.9	0.8	0.75
2	A	158.6	2.0	0.4	0.68
3	A	269.0	8.2	0.2	0.75
4	A	304.2	15.4	1.9	0.19
5	A	345.7	20.2	0.9	0.75
6	A	373.9	63.4	0.8	0.65
7	A	431.0	16.5	2.9	0.32
8	A	491.6	0.7	11.8	0.23
9	A	540.6	18.6	6.4	0.08
10	A	601.0	49.2	4.9	0.67
11	A	783.6	24.7	7.2	0.51
12	A	808.1	51.8	5.1	0.36
13	A	906.2	233.4	0.7	0.36
14	A	917.9	32.8	4.2	0.49
15	A	946.3	34.5	9.2	0.50
16	A	1020.0	33.1	2.0	0.39
17	A	1042.2	10.8	5.2	0.64
18	A	1091.7	94.1	0.9	0.75
19	A	1189.2	8.9	10.7	0.14
20	A	1209.3	8.7	2.0	0.50
21	A	1275.6	2.6	2.2	0.73
22	A	1329.9	11.1	1.0	0.72
23	A	1353.9	4.3	1.0	0.53
24	A	1489.5	2.6	4.2	0.59
25	A	1505.9	4.4	2.0	0.69
26	A	1625.1	17.3	6.9	0.74
27	A	1656.4	18.3	5.3	0.63
28	A	1672.1	14.8	4.0	0.68
29	A	3365.8	11.9	170.0	0.11
30	A	3392.4	2.2	111.5	0.13
31	A	3405.4	6.4	173.3	0.12
32	A	3414.9	7.0	77.9	0.18
33	A	3478.2	3.4	86.9	0.25
34	A	3545.2	3.4	53.0	0.31
35	A	3546.7	7.4	91.4	0.57
36	A	3555.2	8.4	34.3	0.55

15. Dissociation into N<sub>4</sub>H<sub>5</sub> + N<sub>2</sub>H<sub>3</sub> (bond breaking at A).

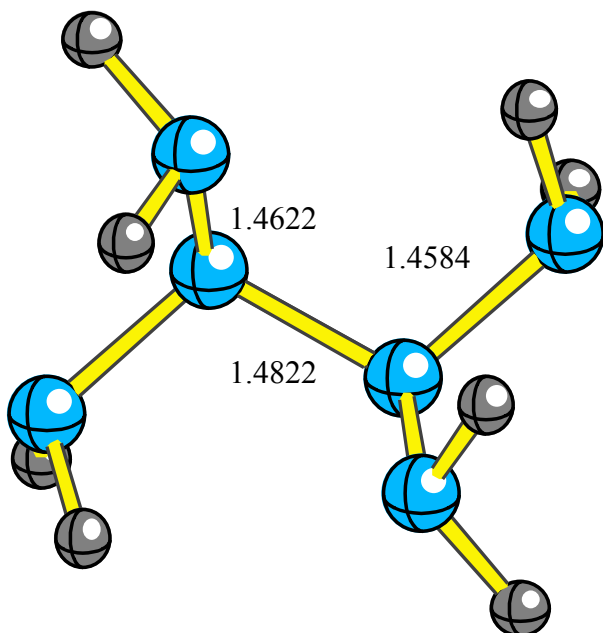
$N_6H_8$ , Hexazine (A)

$C_1$ ,  $^1A$ ,  $H_f=99.92$  kcal/mol,  $I_{sp}=276.01$  sec.



$N_6H_8$ , Tetraaminohydrazine (c,c)

$C_2$ ,  $^1A$ ,  $H_f=84.36$  kcal/mol,  $I_{sp}=253.6$  sec.



Vibrational frequencies of N<sub>6</sub>H<sub>8</sub>, hexazine (A)

No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	61.4	7.3	0.3	0.66
2	A	100.3	9.9	0.6	0.48
3	A	138.4	9.8	0.2	0.54
4	A	295.4	33.9	0.5	0.64
5	A	323.4	6.3	1.1	0.20
6	A	369.5	32.7	0.7	0.71
7	A	434.6	28.7	1.6	0.17
8	A	478.6	18.2	5.3	0.25
9	A	601.1	16.8	1.0	0.59
10	A	748.2	26.3	1.8	0.37
11	A	823.7	67.5	1.8	0.56
12	A	839.6	27.0	5.0	0.75
13	A	910.6	61.0	1.5	0.74
14	A	929.6	98.4	4.5	0.28
15	A	989.6	31.4	10.2	0.05
16	A	1057.4	35.5	7.6	0.04
17	A	1076.8	12.2	4.4	0.52
18	A	1127.9	53.9	3.3	0.50
19	A	1176.4	15.5	2.9	0.45
20	A	1201.3	18.4	3.4	0.38
21	A	1308.1	4.9	2.9	0.75
22	A	1312.1	5.0	1.3	0.75
23	A	1452.5	0.6	11.2	0.29
24	A	1469.1	1.5	3.0	0.70
25	A	1496.3	0.5	3.3	0.71
26	A	1514.4	0.7	2.8	0.75
27	A	1624.6	15.9	6.9	0.75
28	A	1666.3	14.3	7.0	0.58
29	A	3343.0	16.2	171.0	0.10
30	A	3369.0	9.7	141.2	0.52
31	A	3401.3	13.0	164.0	0.12
32	A	3416.4	6.4	231.0	0.09
33	A	3451.3	4.2	163.1	0.21
34	A	3489.3	1.4	100.5	0.16
35	A	3529.9	1.8	51.9	0.62
36	A	3574.8	7.5	75.7	0.57

19. Dissociation into N<sub>5</sub>H<sub>6</sub> + NH<sub>2</sub> (bond breaking at A).

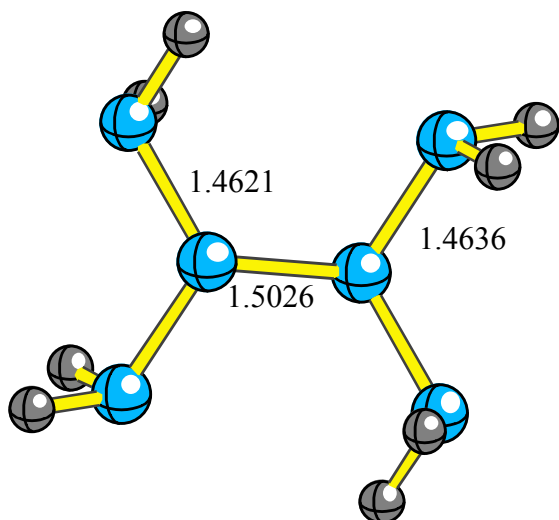
Vibrational frequencies of N<sub>6</sub>H<sub>8</sub>, tetraaminohydrazine (c,c)

No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	139.5	0.8	0.8	0.68
2	B	255.9	59.3	0.3	0.75
3	B	308.3	21.4	0.6	0.75
4	A	309.6	0.8	1.4	0.31
5	A	333.2	66.9	0.6	0.45
6	B	371.0	4.7	1.1	0.75
7	A	399.3	2.8	1.3	0.33
8	A	439.6	16.8	3.2	0.40
9	B	520.0	49.8	0.3	0.75
10	A	597.1	6.1	8.5	0.20
11	B	660.9	6.4	1.2	0.75
12	A	742.2	16.5	16.6	0.06
13	B	845.6	72.5	4.0	0.75
14	A	851.0	35.0	1.8	0.08
15	A	928.6	0.3	6.4	0.07
16	B	950.9	190.4	0.3	0.75
17	B	1075.3	35.2	3.5	0.75
18	A	1123.1	28.4	4.2	0.14
19	B	1181.0	103.0	2.3	0.75
20	A	1203.3	5.4	7.8	0.58
21	B	1268.2	0.9	1.5	0.75
22	A	1276.3	0.5	5.0	0.60
23	A	1354.9	8.7	1.2	0.40
24	B	1369.5	8.2	0.3	0.75
25	B	1624.9	25.5	4.1	0.75
26	A	1632.3	1.5	8.9	0.73
27	B	1639.7	34.0	5.4	0.75
28	A	1649.3	17.5	2.7	0.23
29	B	3374.9	13.7	4.5	0.75
30	A	3375.4	2.1	202.5	0.05
31	B	3416.6	0.3	39.1	0.75
32	A	3422.1	2.6	168.5	0.04
33	B	3531.4	3.7	71.3	0.75
34	A	3532.3	8.6	109.2	0.21
35	A	3555.4	0.6	169.2	0.43
36	B	3555.6	14.7	2.8	0.75

12. Dissociation into 2N<sub>3</sub>H<sub>4</sub>.

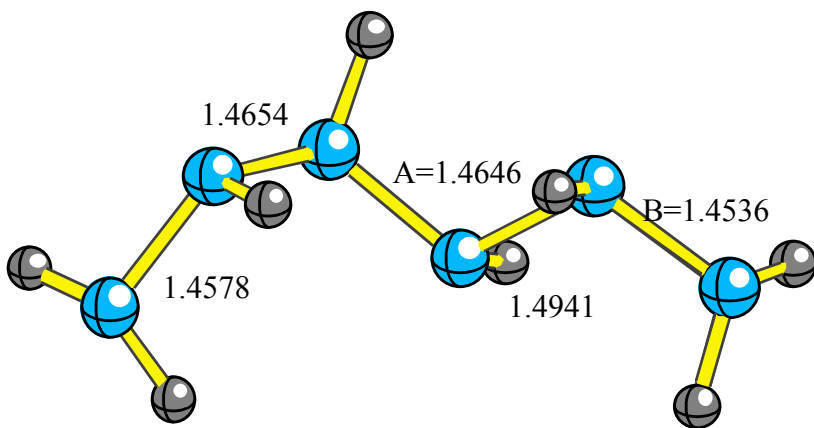
$N_6H_8$ , Tetraaminohydrazine (t,t)

$C_i$ ,  $^1A_g$ ,  $H_f=87.12$  kcal/mol,  $I_{sp}=257.7$  sec.



$N_6H_8$ , Hexazine (B)

$C_1$ ,  $^1A$ ,  $H_f=95.19$  kcal/mol,  $I_{sp}=269.4$  sec.



Vibrational frequencies of N<sub>6</sub>H<sub>8</sub>, tetraaminohydrazine (t,t)

No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	Au	112.7	7.3	0.0	0.00
2	Au	243.5	98.7	0.0	0.00
3	Ag	270.4	0.0	3.8	0.54
4	Au	277.7	19.9	0.0	0.00
5	Ag	285.3	0.0	4.2	0.66
6	Au	318.5	20.0	0.0	0.00
7	Au	426.8	12.5	0.0	0.00
8	Ag	430.7	0.0	6.8	0.23
9	Au	534.1	87.3	0.0	0.00
10	Ag	540.7	0.0	8.3	0.34
11	Ag	599.3	0.0	13.6	0.03
12	Ag	760.4	0.0	21.6	0.68
13	Ag	859.3	0.0	4.8	0.71
14	Au	876.7	80.1	0.0	0.00
15	Au	943.1	264.7	0.0	0.00
16	Ag	953.8	0.0	6.4	0.13
17	Au	1069.5	54.5	0.0	0.00
18	Ag	1103.9	0.0	12.4	0.56
19	Au	1186.8	109.4	0.0	0.00
20	Ag	1218.8	0.0	5.1	0.65
21	Ag	1285.4	0.0	4.4	0.64
22	Au	1294.1	8.8	0.0	0.00
23	Au	1365.8	22.2	0.0	0.00
24	Ag	1382.6	0.0	3.9	0.57
25	Au	1635.6	37.9	0.0	0.00
26	Ag	1638.9	0.0	12.3	0.74
27	Au	1650.3	37.7	0.0	0.00
28	Ag	1650.9	0.0	8.2	0.74
29	Au	3337.6	31.3	0.0	0.00
30	Ag	3343.7	0.0	335.0	0.12
31	Au	3376.1	15.6	0.0	0.00
32	Ag	3377.2	0.0	262.4	0.18
33	Au	3542.9	16.2	0.0	0.00
34	Ag	3544.6	0.0	112.8	0.40
35	Ag	3554.6	0.0	156.0	0.40
36	Au	3554.7	21.0	0.0	0.00

12. Dissociation into 2N<sub>3</sub>H<sub>4</sub>.

13. Dissociation into 2N<sub>3</sub>H<sub>4</sub>.

Vibrational frequencies of N<sub>6</sub>H<sub>8</sub>, hexazine (B)

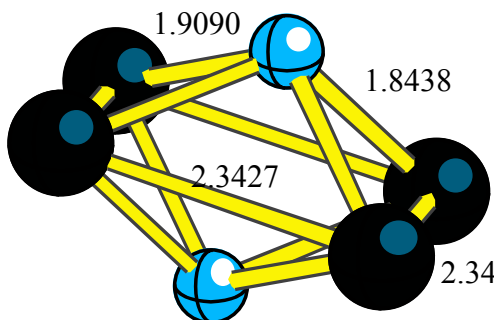
No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	35.1	4.1	0.7	0.73
2	A	85.9	2.0	0.8	0.54
3	A	148.3	5.4	0.4	0.74
4	A	291.2	21.5	0.8	0.10
5	A	339.9	11.8	2.4	0.12
6	A	397.0	14.3	7.0	0.25
7	A	428.1	35.8	0.8	0.39
8	A	543.2	35.7	5.4	0.33
9	A	585.7	49.7	2.7	0.29
10	A	749.3	9.6	4.8	0.73
11	A	796.7	74.2	3.6	0.73
12	A	867.2	139.8	0.5	0.73
13	A	894.3	38.2	12.4	0.57
14	A	932.9	115.5	0.6	0.20
15	A	972.5	32.2	1.7	0.69
16	A	1040.1	31.1	2.0	0.67
17	A	1081.2	26.3	4.8	0.05
18	A	1117.1	15.1	6.9	0.08
19	A	1191.4	19.7	2.4	0.65
20	A	1202.7	6.9	6.4	0.06
21	A	1304.8	4.7	1.9	0.52
22	A	1325.2	2.5	1.3	0.62
23	A	1441.6	6.5	1.3	0.74
24	A	1465.1	5.9	0.7	0.58
25	A	1500.7	0.7	2.0	0.67
26	A	1535.9	2.2	5.6	0.74
27	A	1654.6	1.1	5.2	0.70
28	A	1657.8	30.1	4.6	0.61
29	A	3404.9	9.1	131.3	0.10
30	A	3414.0	5.0	116.8	0.06
31	A	3424.2	1.9	194.2	0.10
32	A	3442.1	2.4	42.0	0.35
33	A	3447.4	1.2	162.4	0.24
34	A	3470.5	5.0	15.6	0.30
35	A	3538.5	3.7	62.4	0.51
36	A	3546.5	5.6	56.3	0.60

13. Dissociation into N<sub>3</sub>H<sub>4</sub> + N<sub>2</sub>H<sub>2</sub> + NH<sub>2</sub> (two bond breaking at A and B).

**Lithium substituted hydrazine derivatives:  $N_xLi_y$  molecules.**

$N_2Li_4$

$C_{2h}$ ,  $^1A$ ,  $H_f=79.83$  kcal/mol,  $I_{sp}=317.03$  sec.



Initial geometry for geometry optimization was hydrazine type structure.

No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	Au	219.1	0.0	0.0	0.41
2	Au	375.3	40.7	0.0	0.46
3	Bu	376.0	40.8	0.0	0.75
4	Ag	472.8	0.0	973.4	0.00
5	Bg	484.5	0.0	196.0	0.75
6	Ag	485.3	0.0	149.2	0.75
7	Ag	486.3	0.0	58.0	0.73
8	Bu	514.5	299.0	0.0	0.75
9	Bg	597.6	0.0	143.9	0.75
10	Au	727.6	214.8	0.0	0.04
11	Bu	728.0	214.9	0.0	0.62
12	Ag	735.1	0.0	325.9	0.03

2.Li-Li-Li-Li ring opening.

3.Li-Li-Li-Li ring opening.

7. Li ring rupture (two bonds breaking).

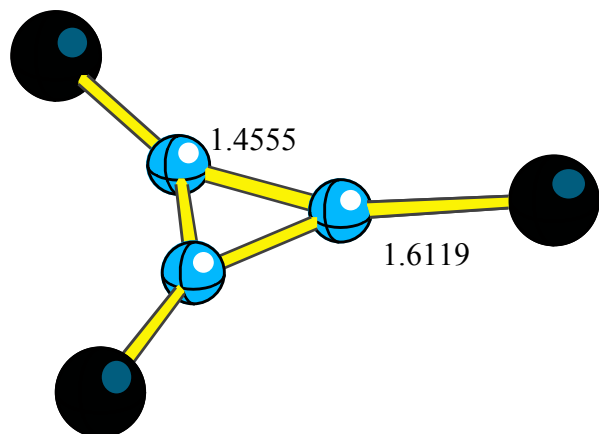
10. Li-N bond breaking (two bonds breaking).

11.Li-N bond breaking (four bonds breaking).

12. Dissociation into  $2Li + N_2Li_2$ .



$C_{3h}$ ,  $^1A'$ ,  $H_f=173.50$  kcal/mol,  $I_{sp}=440.3$  sec.



No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	E'	62.8	54.1	913.8	0.74
2	A'	155.7	0.0	0.0	0.27
3	A''	158.0	60.0	10.0	0.75
4	E''	253.3	0.0	329.1	0.75
5	E'	560.5	81.7	> 1000	0.72
6	A'	638.2	0.0	> 1000	0.19
7	E'	848.5	23.4	> 1000	0.70
8	A'	1352.5	0.0	> 1000	0.17

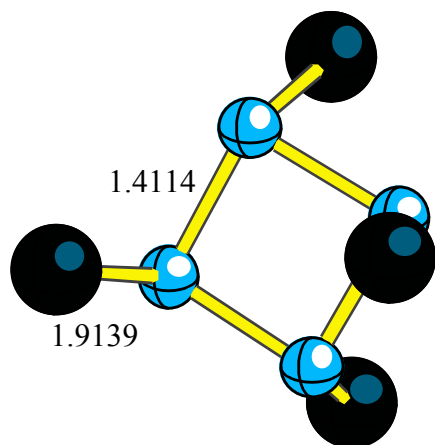
5. Ring opening.

7. Dissociation into Li +  $N_3Li_2$ .

8. Dissociation into 3NLi (three bonds breaking).

$N_4Li_4$

$S_4$ ,  $^1A$ ,  $H_f=138.48$  kcal/mol,  $I_{sp}=340.7$  sec.



No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	34.5	10.1	60.3	0.75
2	A	34.5	10.1	60.3	0.75
3	A	53.9	2.7	409.1	0.75
4	A	55.5	0.0	206.0	0.00
5	A	108.8	1.8	389.8	0.75
6	A	108.8	1.8	389.8	0.75
7	A	236.2	0.0	> 1000	0.00
8	A	262.7	59.5	632.5	0.75
9	A	321.8	0.0	962.1	0.07
10	A	572.2	48.4	114.4	0.75
11	A	575.0	11.8	770.0	0.75
12	A	575.0	11.8	770.0	0.75
13	A	794.0	0.0	> 1000	0.10
14	A	878.6	5.1	9.5	0.75
15	A	878.6	5.1	9.5	0.75
16	A	1142.3	7.2	5.2	0.75
17	A	1143.7	0.0	801.2	0.01
18	A	1212.9	0.5	12.2	0.75

14. Dissociation into  $NLi + N3Li_3$  (two N-N bonds breaking).

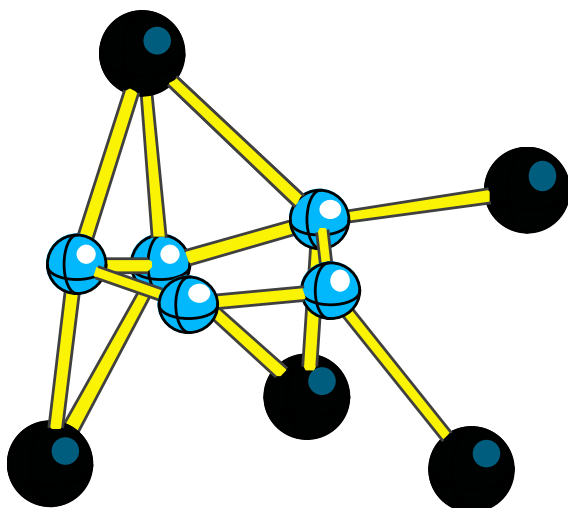
15. Dissociation into  $NLi + N3Li_3$  (two N-N bonds breaking).

17. Dissociation into  $4NLi$  (four N-N bonds breaking).

18. Dissociation into  $2N_2Li_2$  (two N-N bonds breaking).

$N_5Li_5$

$C_1$ ,  $^1A$ ,  $H_f=141.11$  kcal/mol,  $I_{sp}=307.6$  sec.



No.	Symm.	Frequency	IR Intensity	Raman Intensity	Depolarization Ratio
1	A	81.0	24.6	638.5	0.20
2	A	101.9	9.5	582.2	0.26
3	A	145.4	6.9	80.5	0.51
4	A	169.7	13.6	> 1000	0.16
5	A	204.0	41.2	988.3	0.40
6	A	210.8	9.9	75.8	0.25
7	A	260.8	25.3	> 1000	0.10
8	A	273.5	15.0	> 1000	0.05
9	A	311.2	9.0	935.9	0.12
10	A	343.7	86.6	300.6	0.74
11	A	406.5	5.0	416.1	0.57
12	A	422.2	4.7	> 1000	0.15
13	A	490.4	32.5	430.9	0.18
14	A	504.7	3.7	> 1000	0.07
15	A	532.0	0.8	> 1000	0.73
16	A	607.1	32.5	114.1	0.23
17	A	662.3	101.1	> 1000	0.23
18	A	716.2	3.4	146.1	0.72
19	A	741.2	4.1	778.7	0.19
20	A	848.6	6.0	100.2	0.32
21	A	920.0	8.9	161.6	0.14
22	A	1032.7	4.3	47.2	0.67
23	A	1069.2	36.9	458.8	0.15
24	A	1354.7	39.3	271.6	0.12

15,16,17,18,22,23,24. Ring opening.

19,21. Ring opening (two bonds breaking).

## Conclusions.

1. Currently used propellants have calculated  $I_{sp}$  less than 200 sec, and the lowest dissociation mode vibrational frequency at more than  $800\text{ cm}^{-1}$ .
2. Hydrazine derivatives, especially ring compounds, have  $I_{sp}$  about 300 to 360 sec, thus have potential as HEDM molecules. The lowest dissociation mode vibrational frequency is at the similar range as the currently used molecules. Also, as ring compounds must involve two bond breaking for dissociation into two fragments, the activation barrier is expected to be higher than linear molecules of similar kind.
3. Lithium substitution on hydrogen position further increases  $I_{sp}$  without lowering of the lowest dissociation frequency. For example,  $I_{sp}=440\text{ sec}$  for  $\text{N}_3\text{Li}_3$ .

## Acknowledgements

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